

Using Mathematical Modeling to Identify Causes of Souring During Food Waste Anaerobic Co-Digestion

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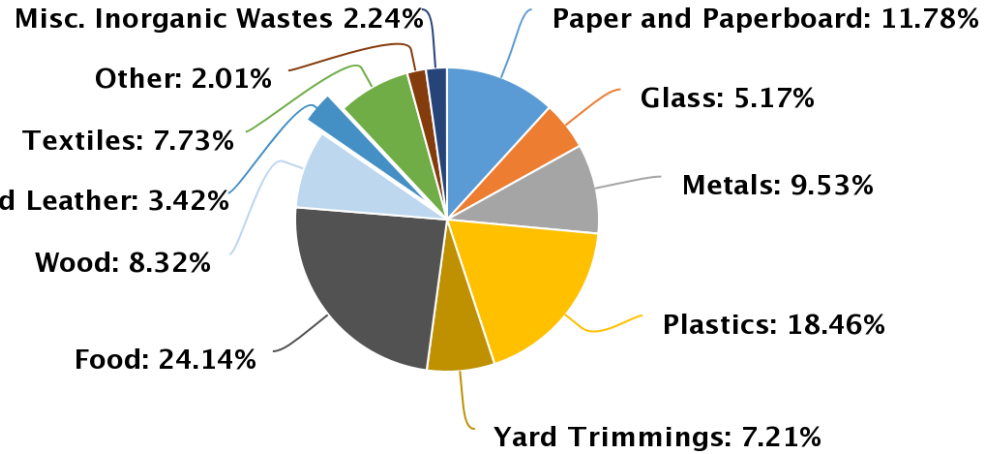
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2022

Total MSW Landfill by Material, 2018

146.1 million tons



24% of U.S. municipal solid waste is food waste (FW)¹



¹U.S. EPA. 2022. National Overview: Facts and Figures on Materials, Wastes and Recycling



6-8% of global greenhouse gas emissions as associated with FW³

³U.S. EPA. (2018). Advanced Sustainable Materials Fact Sheet.
www.nbcnews.com/mach/science/simple-way-we-might-turn-food-waste-green-energy-ncna827166

6 states have banned landfilling of FW⁴



⁴<https://www.rts.com/resources/guides/food-waste-America>
baltimoresun.com/maryland/baltimore-county/bs-md-co-eastern-sanitary-landfill

~1600 anaerobic digestion facilities in the US⁶

⁶American biogas council, <https://americanbiogascouncil.org/biogas-market-snapshot>
Picture: www.ennead.com



Only 141 facilities perform FW digestion or co-digestion⁷

⁷US EPA. Anaerobic Digestion Facilities Processing Food Waste in the United States (2017 & 2018)



Why are so few plants performing FW co-digestion?

FW co-digestion start up can be difficult

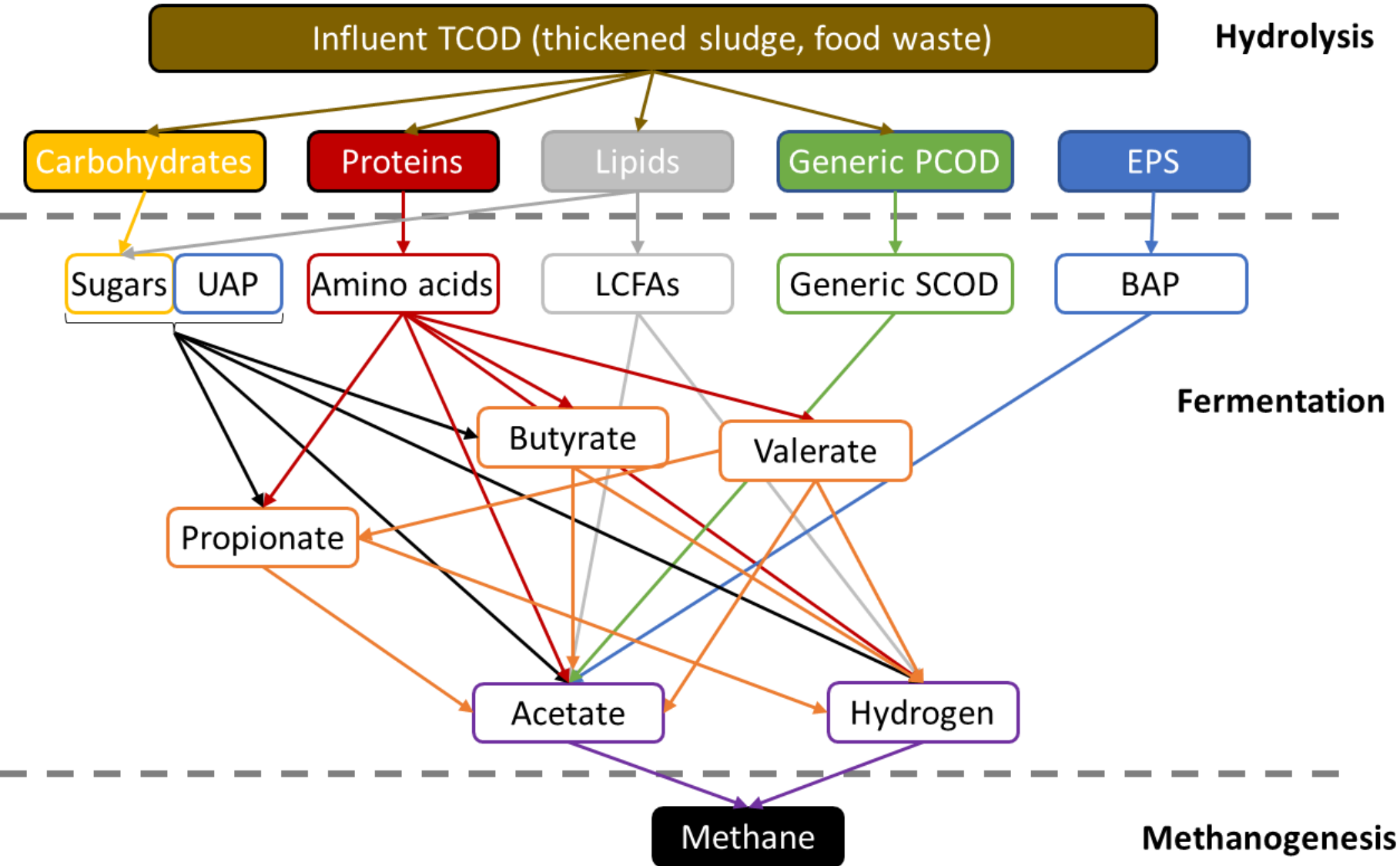
- Food waste feed stream variability⁸
 - Carbohydrates (6-48% DW)
 - Proteins (19-60% DW)
 - Lipids (11-36% DW)
- C/N ratio
 - 20-45 FW⁹ vs. 6-12 AD sludge
- Inhibition from volatile fatty acids (VFAs)
- Souring (pH < 6.0)

⁸Morales-Polo et al. (2018). *Appl. Sci.* 8:1804; ⁹Chiu & Lo. (2016). *Environ. Sci. Pollut. Res.* 23:24435–24450

Research objective

- Developed Food Waste co-Digestion Model (FWcoDM)
 - Anaerobic Digestion Model 1 (ADM1)
 - Combined Activated Sludge-Anaerobic Digestion Model (CASADM)
- Identify the causes and leading indicators of souring
 - Varying organic loading
 - Varying hydraulic retention time (HRT)
 - Feeding frequency

Features of FWcoDM

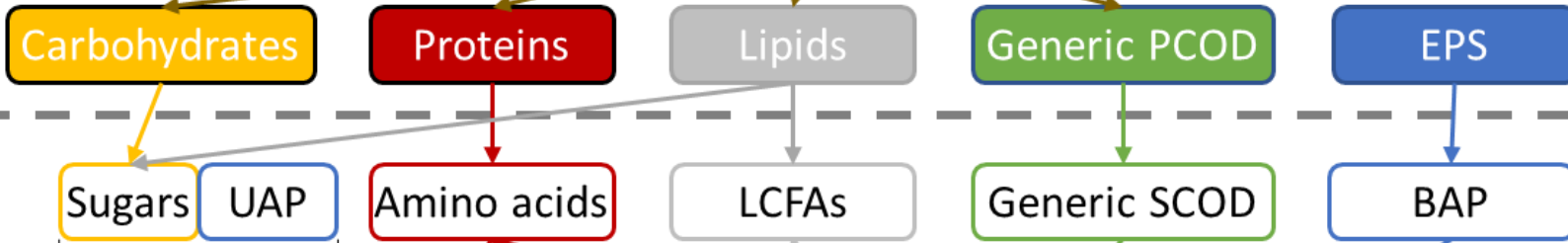


- PCOD=particulate COD (thickened sludge)
- LCFA=long chain fatty acids (palmitate)
- EPS=extracellular polymeric substances
- SMPs=Soluble microbial products (BAP+UAP)
- BAP=biomass associated products
- UAP=utilization associated products

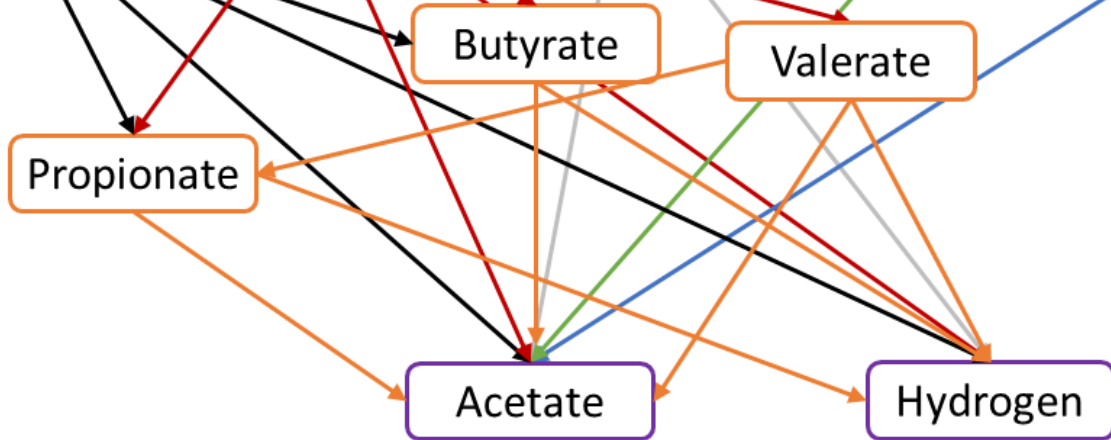
Features of FWcoDM

Influent TCOD (thickened sludge, food waste)

Hydrolysis



Fermentation

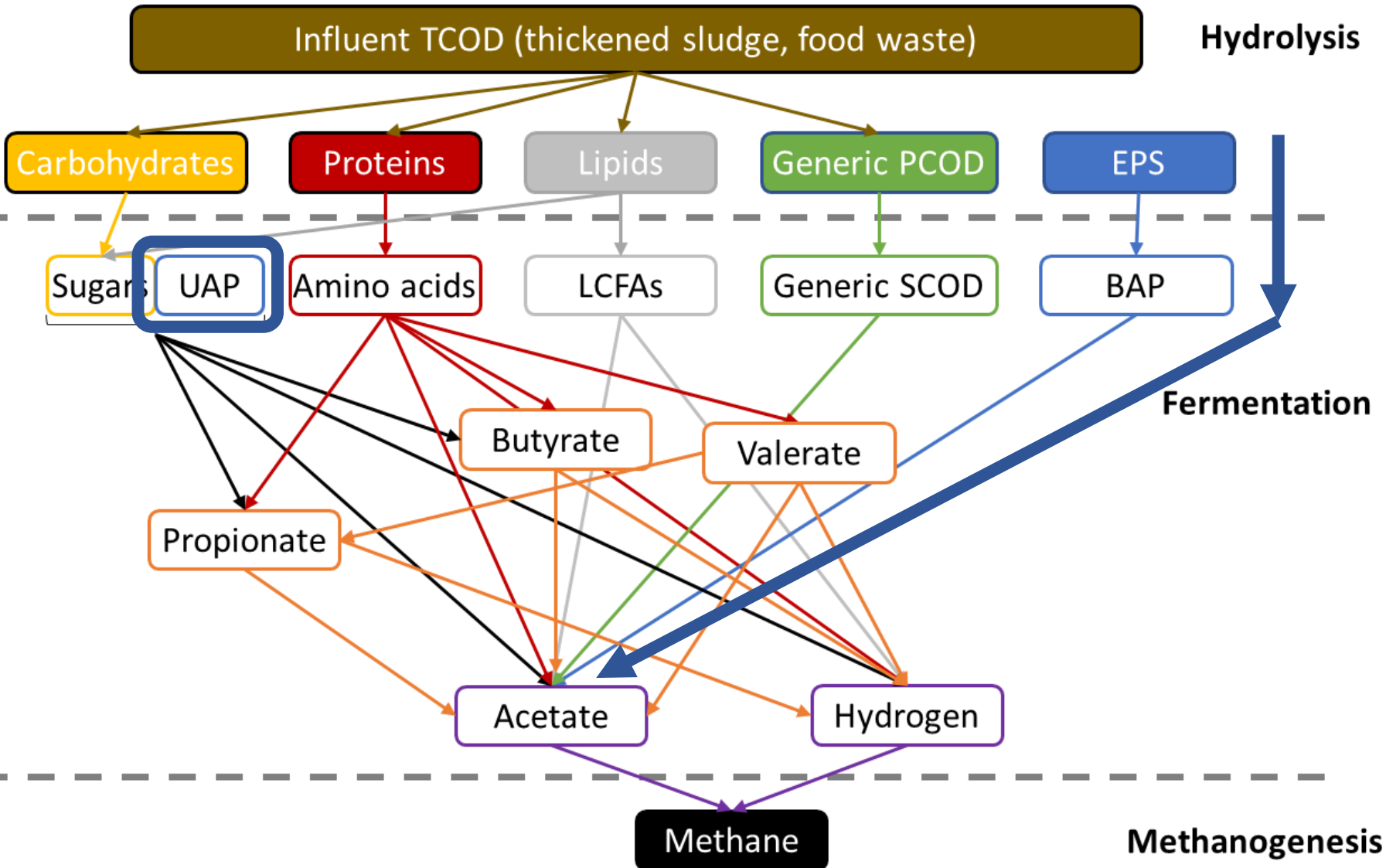


Methanogenesis



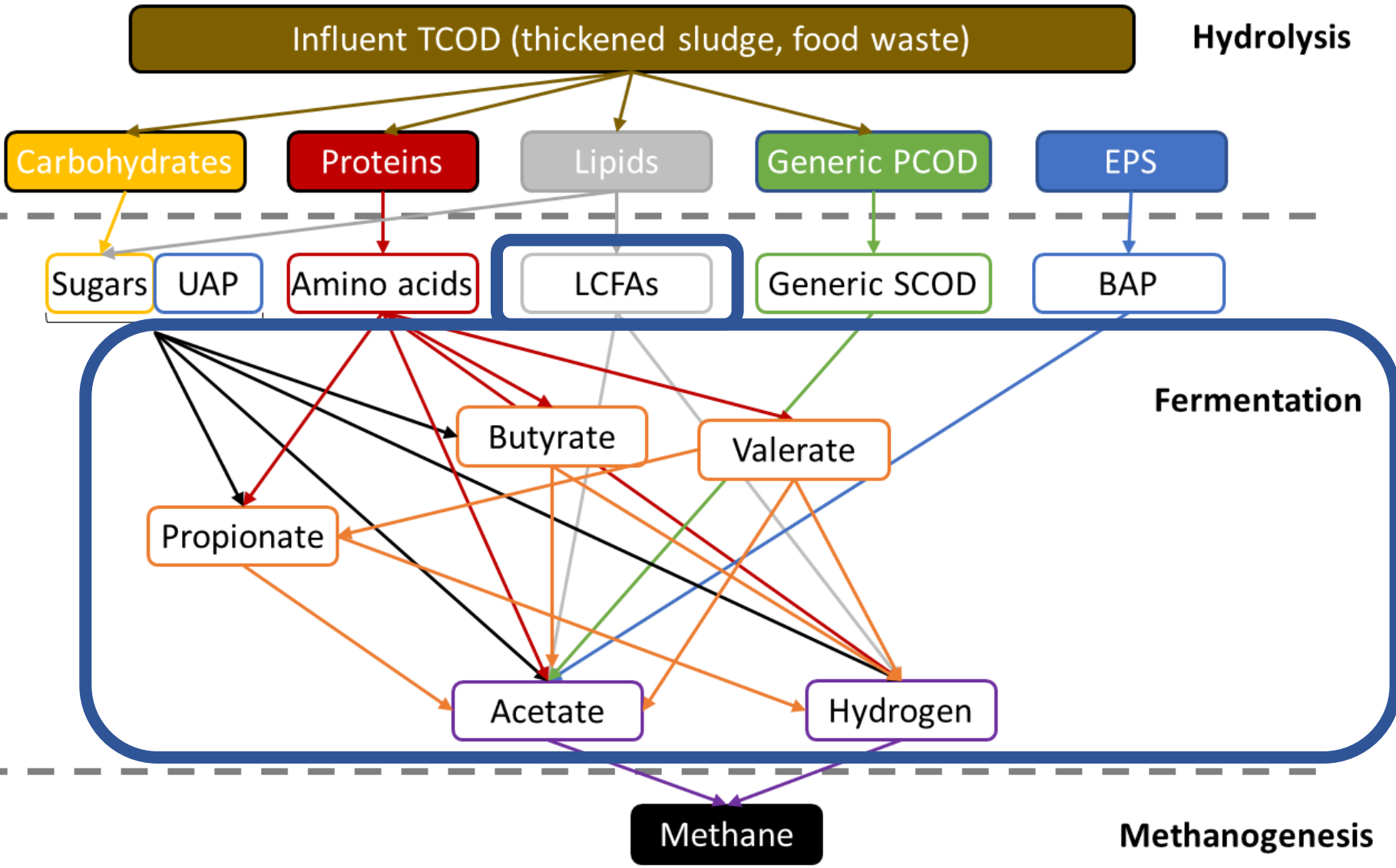
- First order hydrolysis kinetics
- Dual limitation Monod kinetics

Features of FWcoDM



- First order hydrolysis kinetics
- Dual limitation Monod kinetics
- Incorporation of EPS and soluble microbial products (SMPs)

Features of FWcoDM



- First order hydrolysis kinetics
- Dual limitation Monod kinetics
- Incorporation of EPS and soluble microbial products (SMPs)
- Extensive volatile fatty acid (VFA) and LCFA modeling

Inhibition kinetics

- Fermentation

$$I_i = \frac{K_i}{K_i + C_i}$$

- Chemical i (acetate, H₂, LCFAs)
- K = inhibition concentration
- C = chemical concentration

- Methanogenesis

$$I_{\text{pH}} = \exp \left[-3 \left(\frac{\text{pH} - \text{pH}_{\text{UL}}}{\text{pH}_{\text{UL}} - \text{pH}_{\text{LL}}} \right)^2 \right], \text{pH} < \text{pH}_{\text{UL}}$$

$$I_{\text{pH}} = 1, \text{pH} > \text{pH}_{\text{UL}}$$

- UL = upper pH limit
- LL = lower pH limit

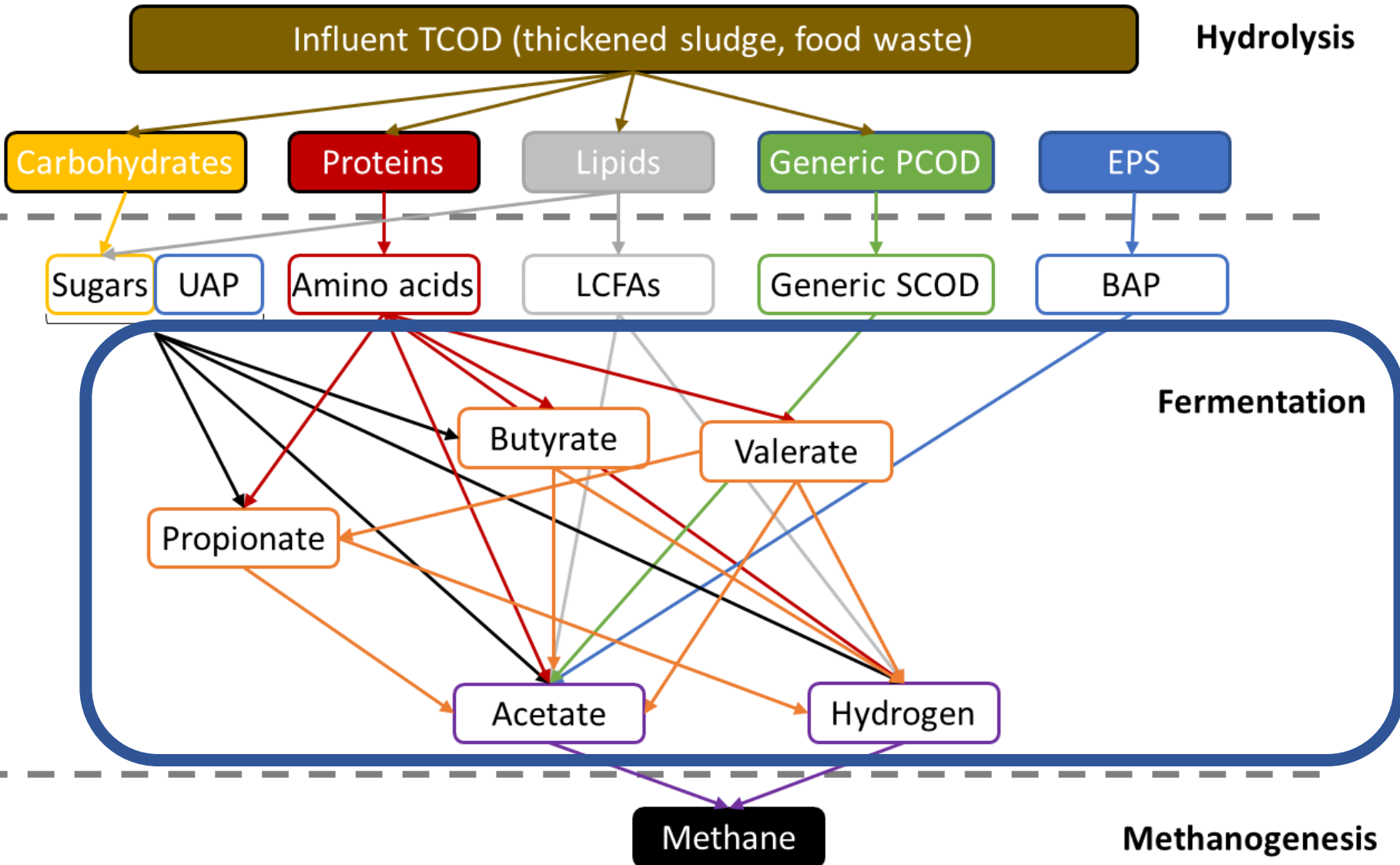
Gas-liquid mass transfer rate

- Mass transfer rate

$$R_i = \frac{V_L}{V_G} K_L a_i (C_i^L - C_i^L H_i RT)$$

- Chemical i (CH₄, CO₂, NH₃, H₂)
- L = liquid, G = gas
- V = Volume
- C^L = Liquid phase concentration
- H = Henry's law constant
- T = Temperature
- R = Gas constant
- K_La = Mass transfer coefficient

Features of FWcoDM



- First order hydrolysis kinetics
- Dual limitation Monod kinetics
- Incorporation of EPS and soluble microbial products (SMPs)
- Extensive fatty acid modeling
- pH and chemical speciation of VFAs, CO₂, and NH₄⁺ using proton balances

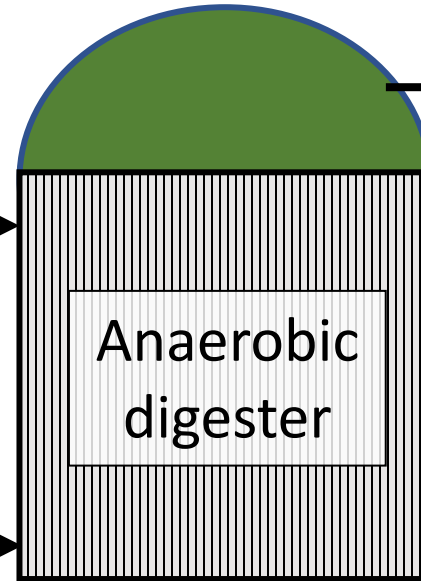
Model parameters

Food waste

- 200 g TCOD/L
- 31 g VSS/L
 - 41% carbs
 - 32% proteins
 - 27% lipids
- pH 5.2
- Sugar 35 g/L
- Amino acids 35 g/L
- Acetate 295 mg/L
- Propionate 219 mg/L
- Butyrate 44 mg/L
- Valerate 51 mg/L

Thickened sludge (ThS)

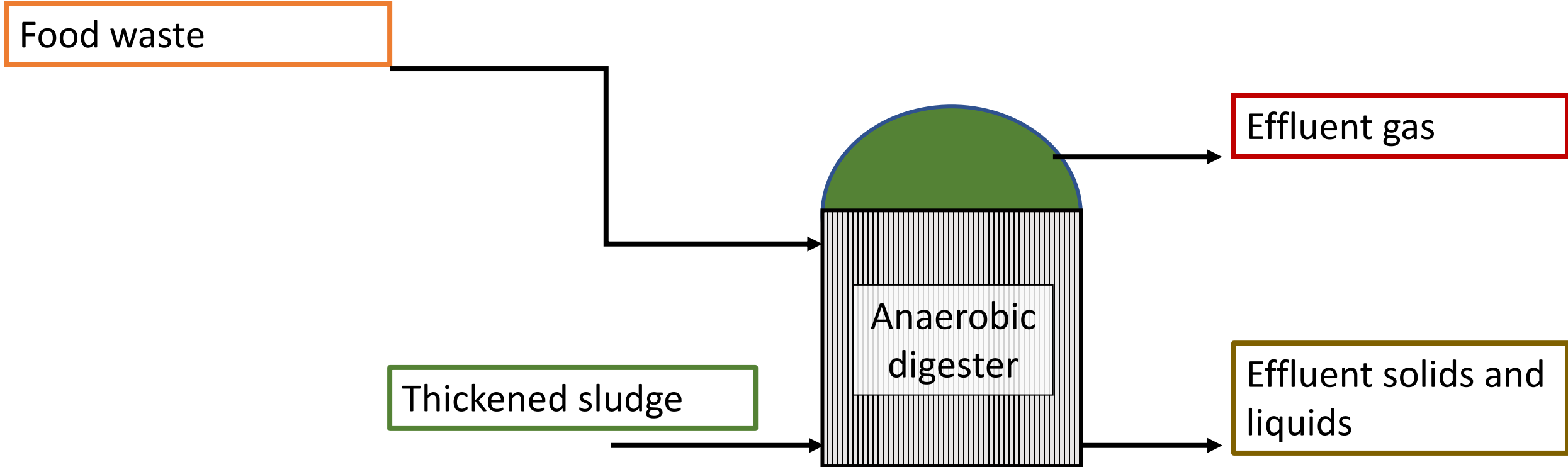
- 50 g TCOD/L
- 1.5 g SCOD/L
- 34 g VSS/L
 - 23% carbs
 - 8% proteins
 - 55% lipids
 - 14% generic PCOD
- pH 6.55



- Gas composition (CH₄, CO₂)
- Biogas production

- pH
- Sludge reduction
- Soluble COD and VFAs
- NH₄⁺
- Bicarbonate alkalinity
- Total alkalinity

Model parameters



$$\text{Vol} \frac{\partial C_h}{\partial t} = \sum_{\substack{\text{ThS} \\ \text{FW}}} Q_{i,\text{in}} C_{i,h,\text{in}} - \sum_{\text{Effluent}} Q_{i,\text{out}} C_{i,h,\text{out}} + \text{Vol} \sum R_h I_{h,\text{pH}}$$

Vol = Reactor volume

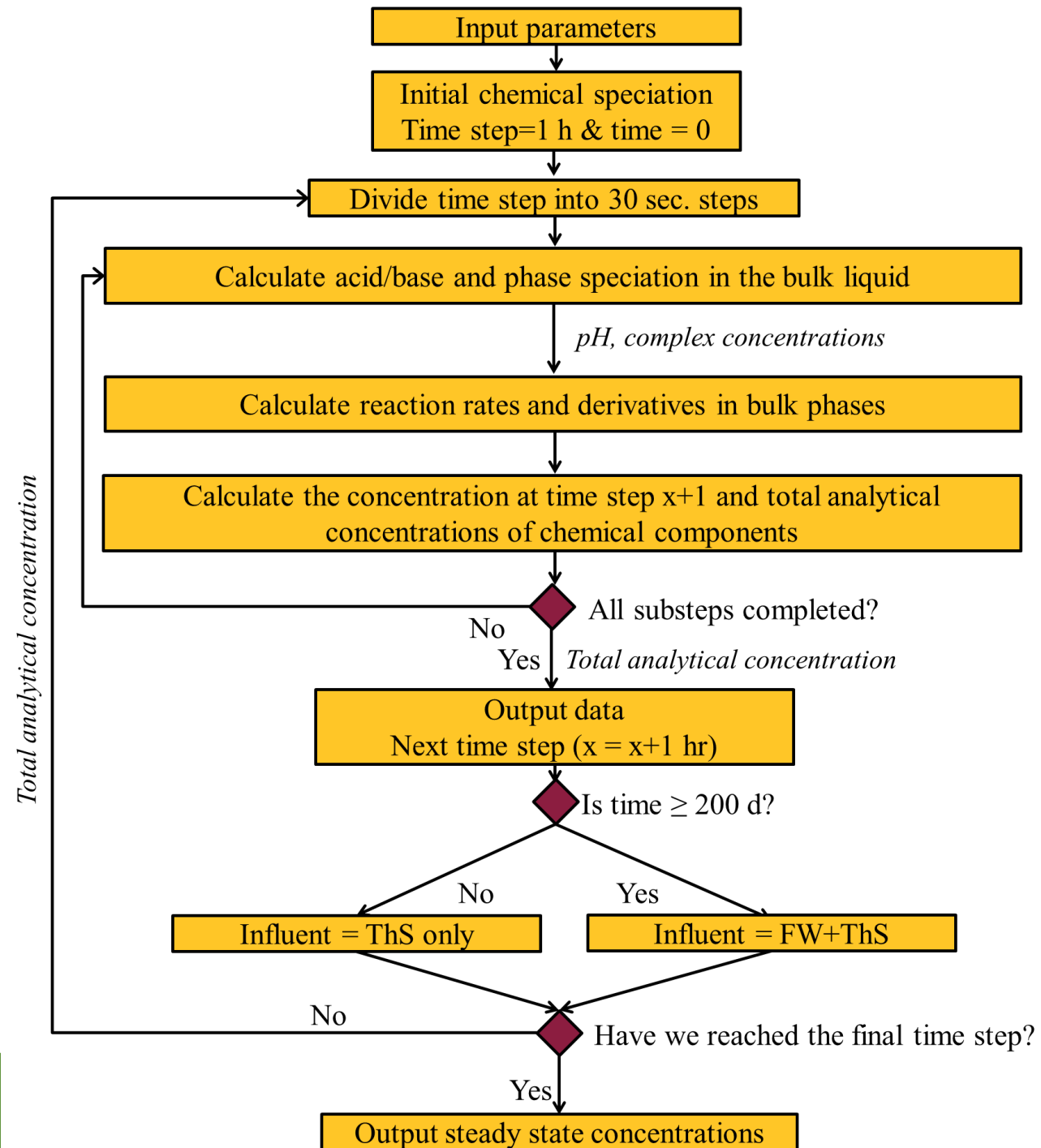
Q = Volumetric flow rate

R = Reaction rate

I = Inhibition rate

Model execution

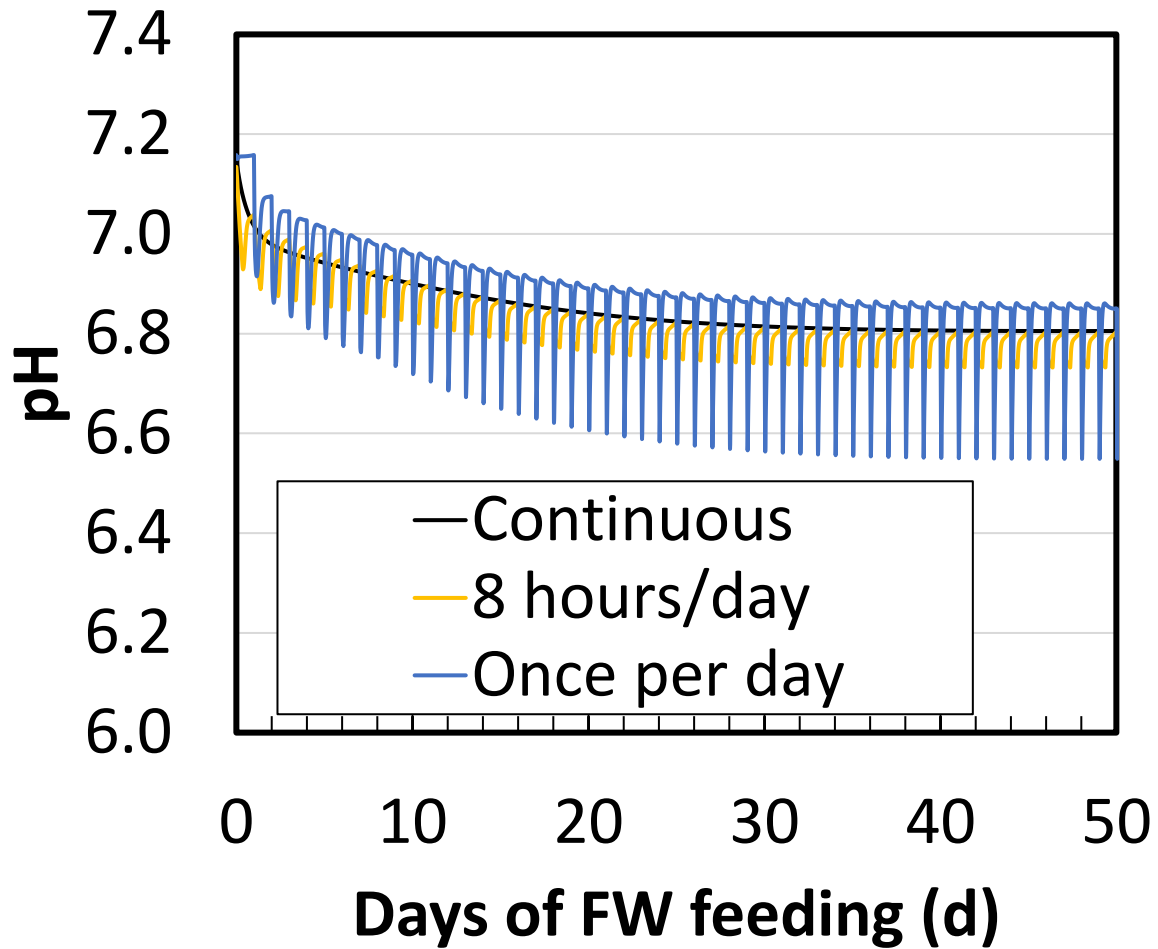
- MATLAB 2021a using ordinary differential equations solver
- Run for 200 d with thickened sludge-only (ThS) as the input
- Run for an additional 200 d with 50% FW+ 50% thickened sludge by volume as input
- $< 10^{-5}$ numerical error
- Souring pH < 6.0
- Total alkalinity = Bicarbonate alkalinity + VFA alkalinity



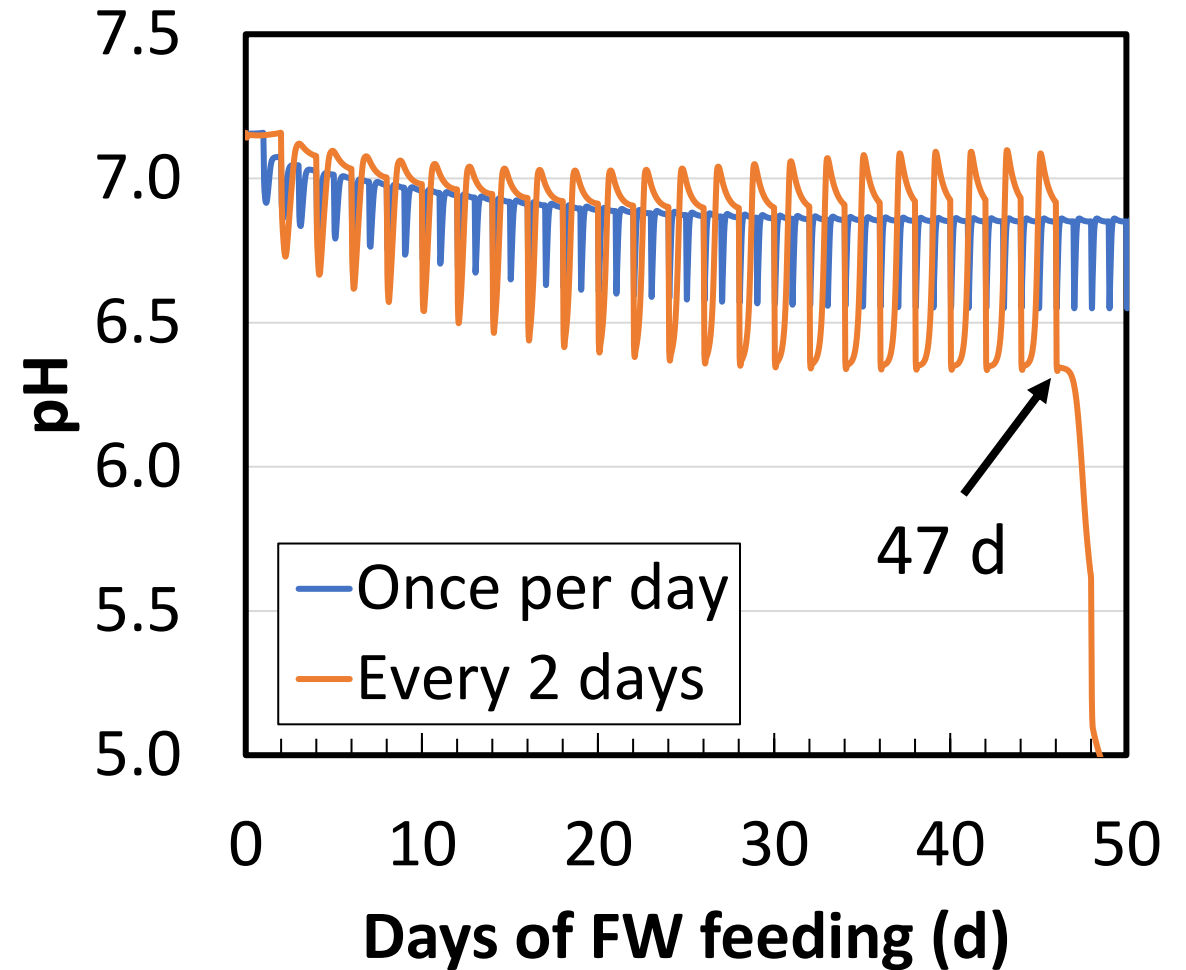
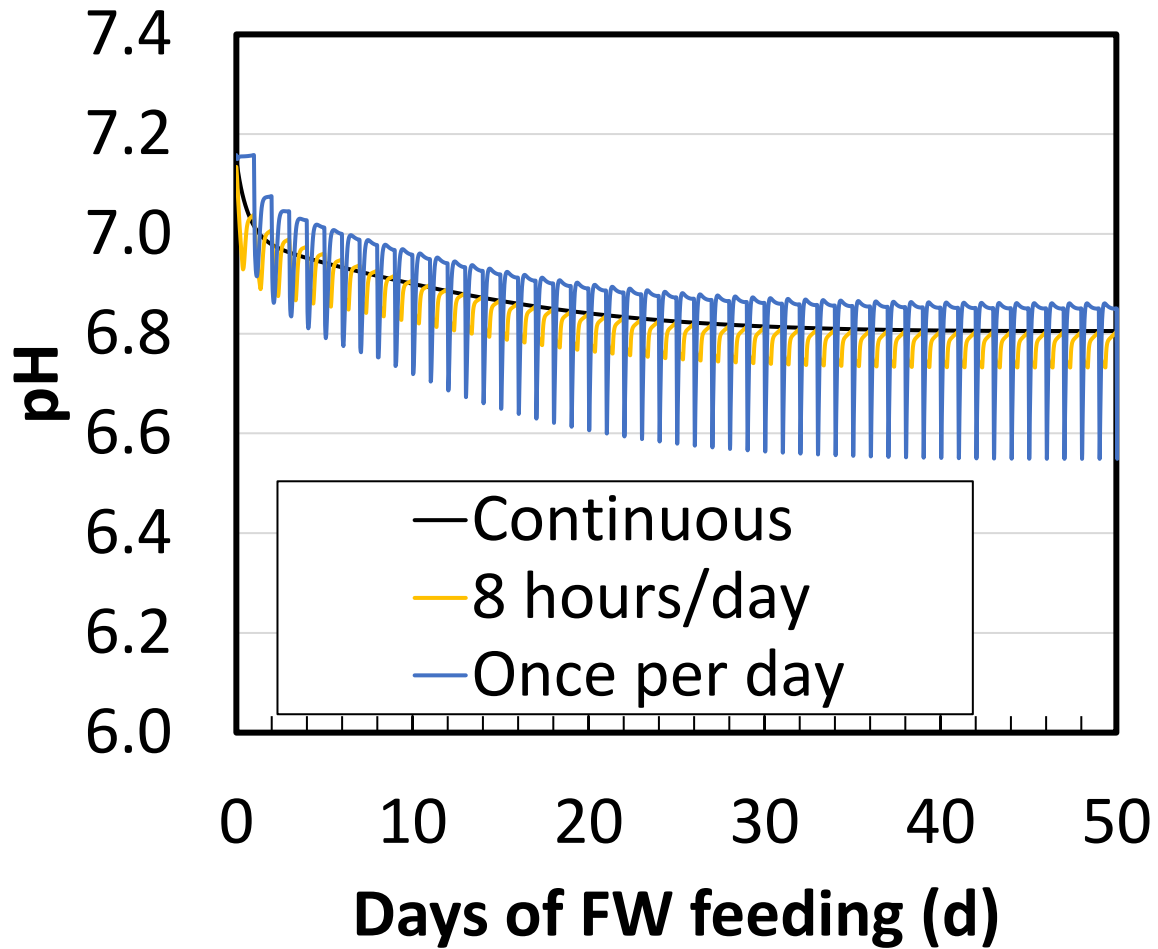
Scenario overview

- Identify the causes and leading indicators of souring at different feeding frequencies
 - Continuous
 - Feed for 8 h/d
 - Feed once daily
 - Feed every 2 days
- Organic loading
 - Every day: 7.0 g/(L-d)
 - Every 2 day: 13.9 g/(L-d)
- 18-d HRT

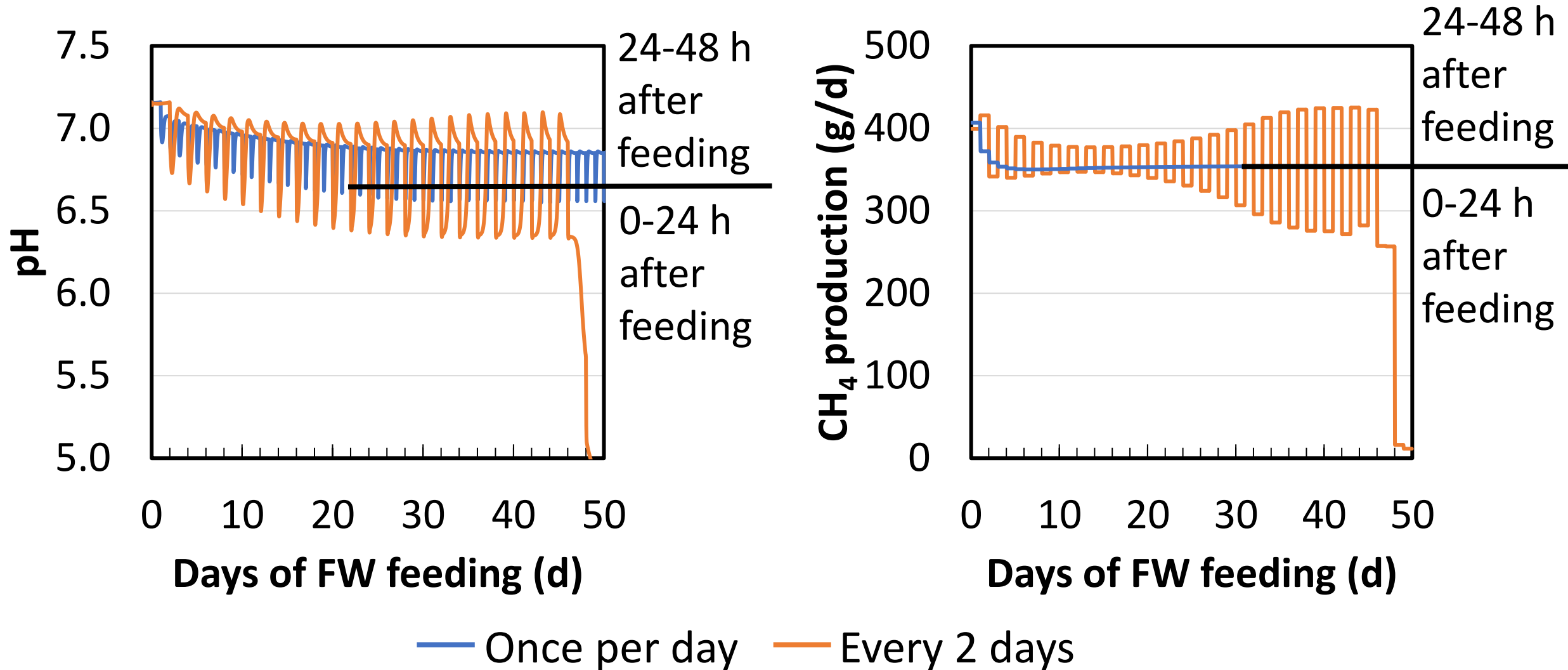
Long-term operations are stable at when fed daily...



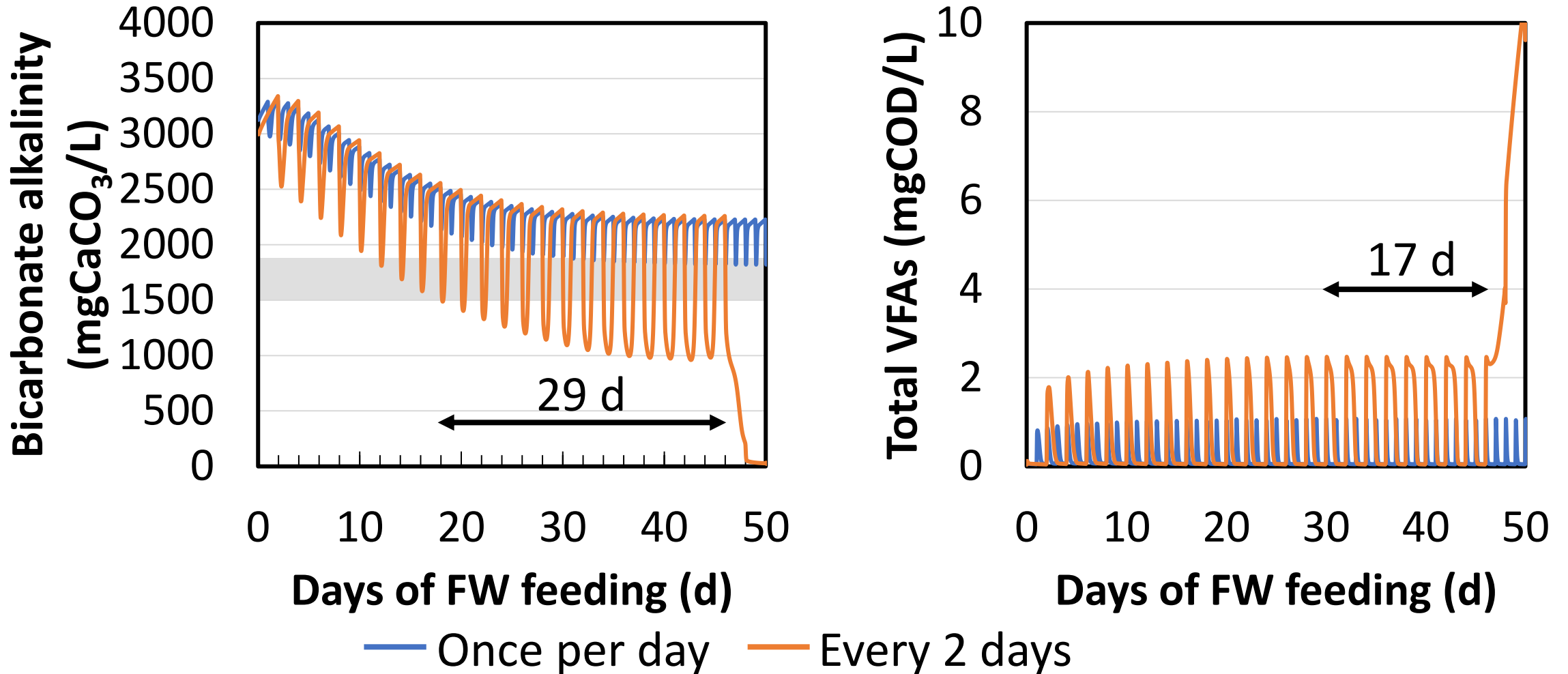
Long-term operations are stable at when fed daily but not when fed every 2 days



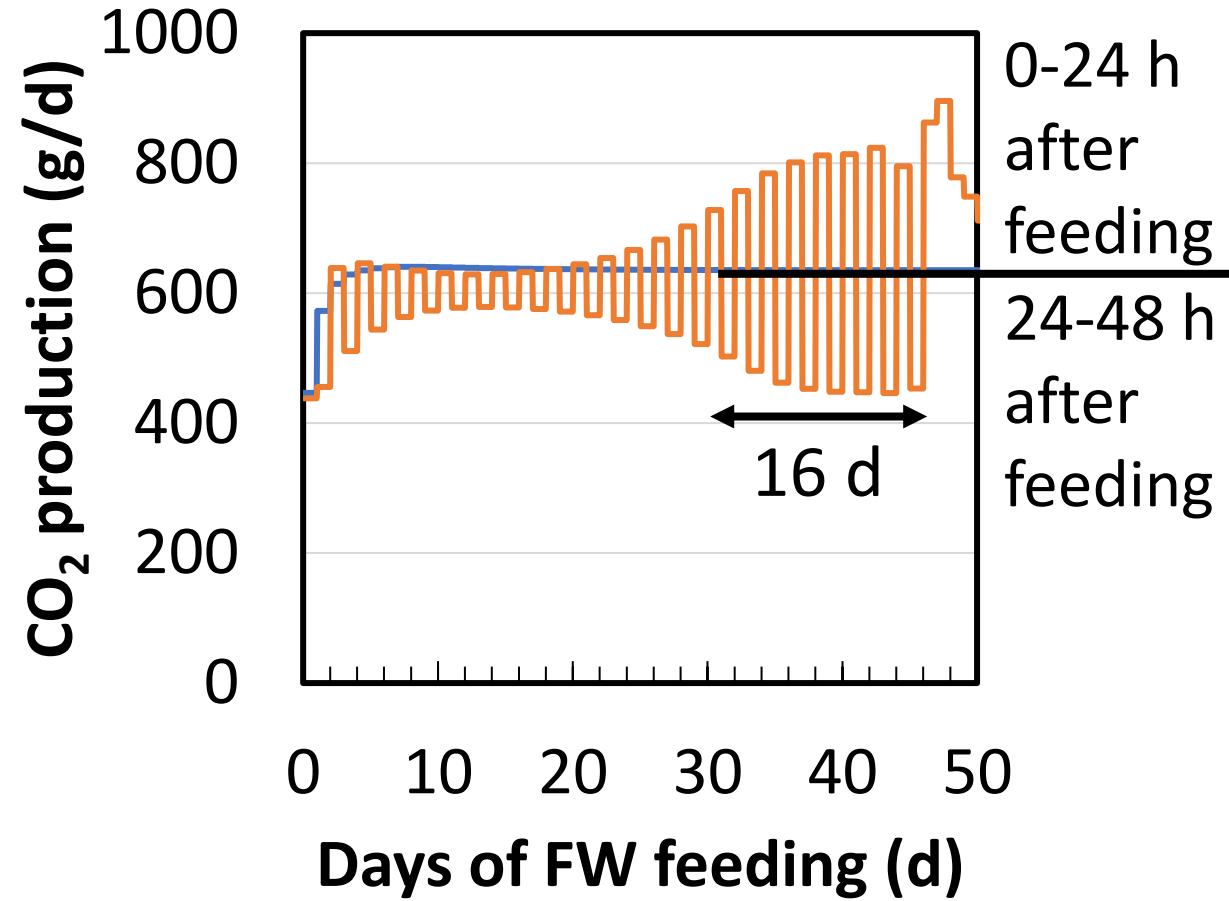
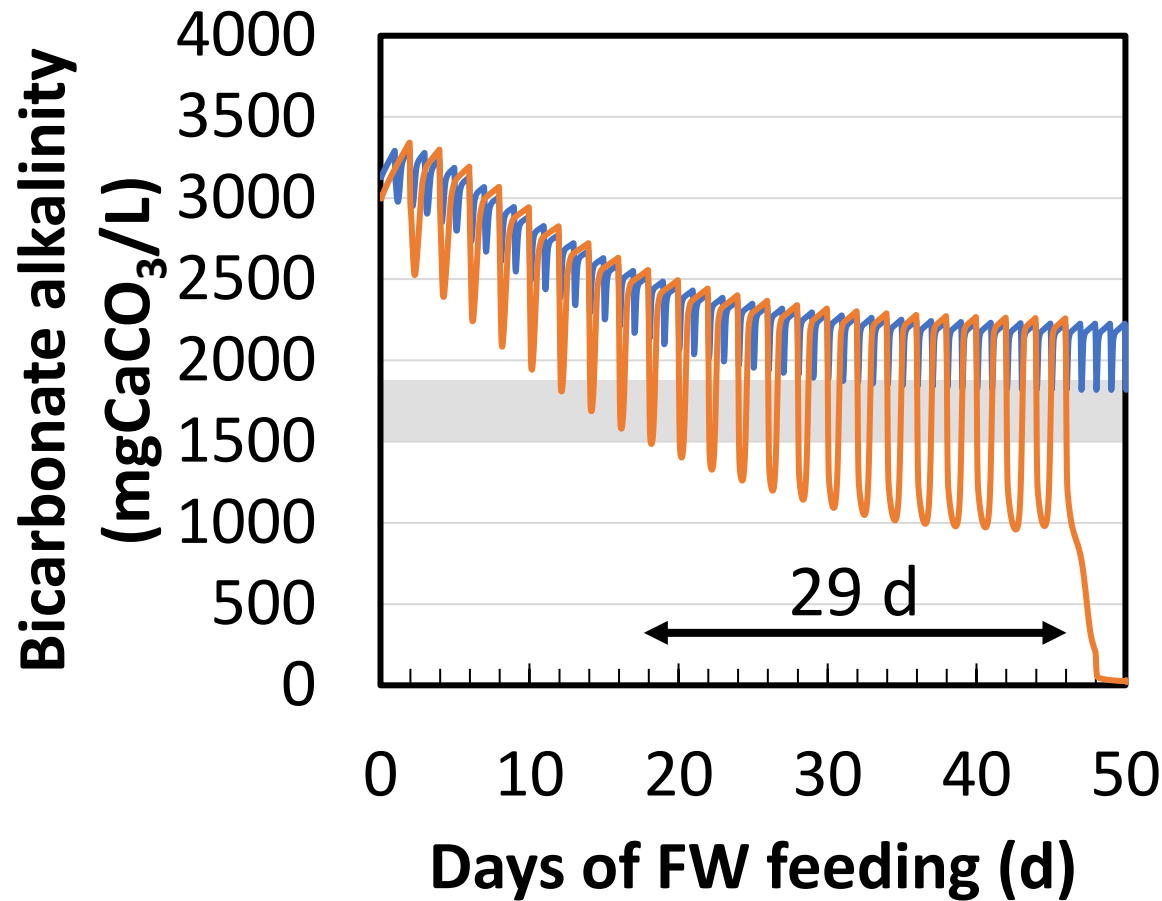
Performance is cyclical



Bicarbonate alkalinity is a longer leading indicator of souring vs. VFA concentration

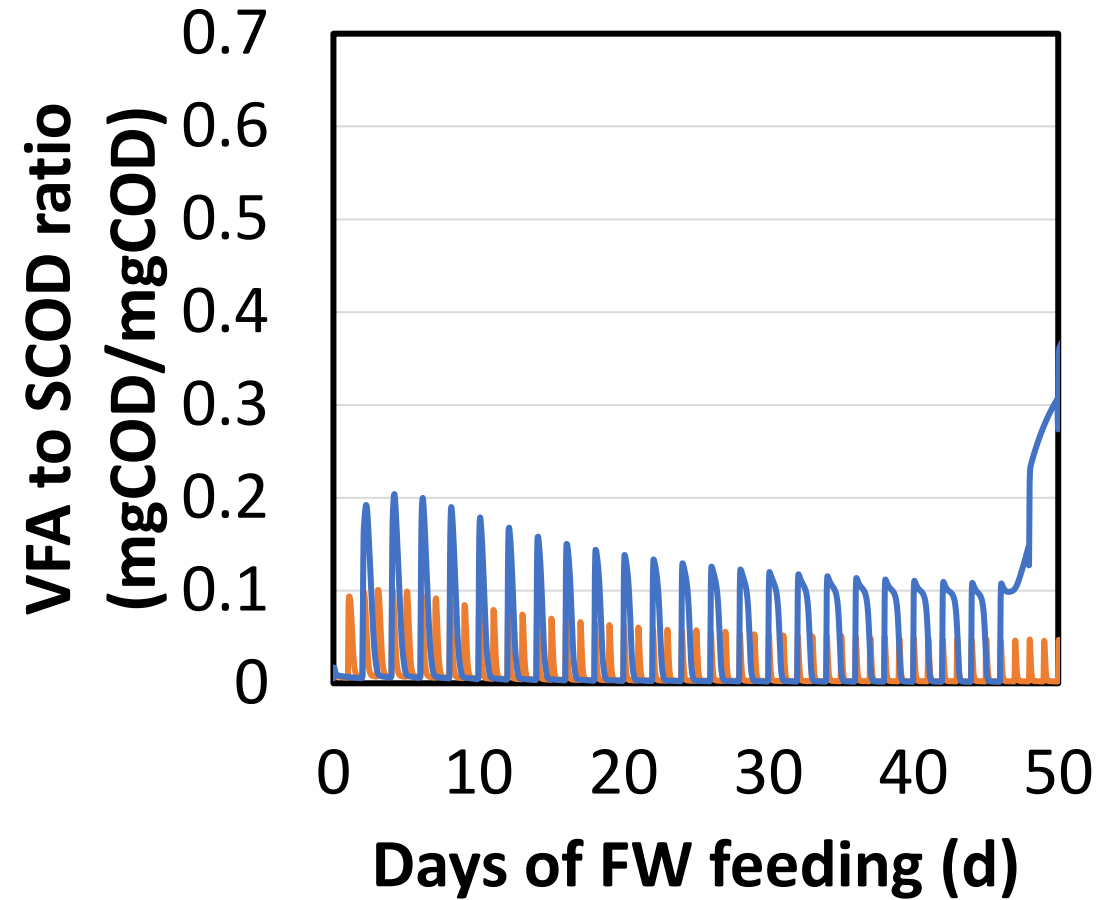
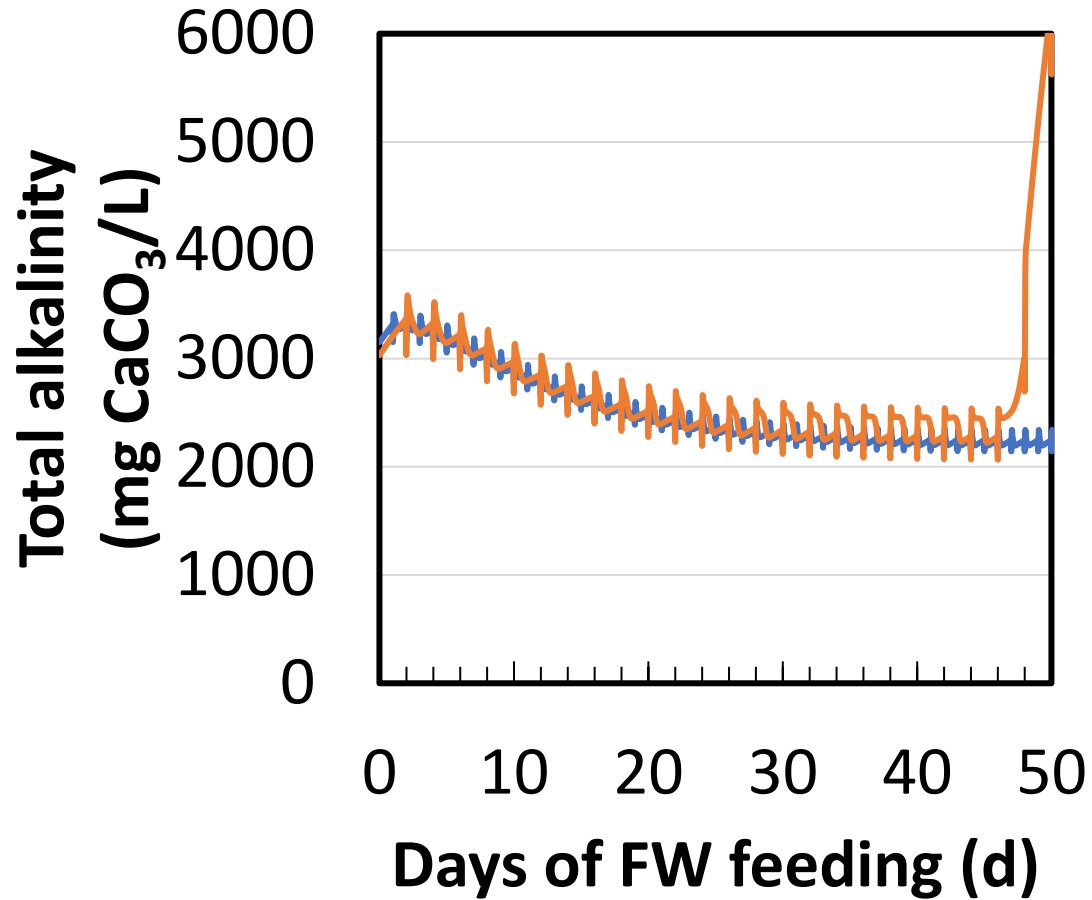


CO₂ off-gassing depleted the system of alkalinity, contributing to souring



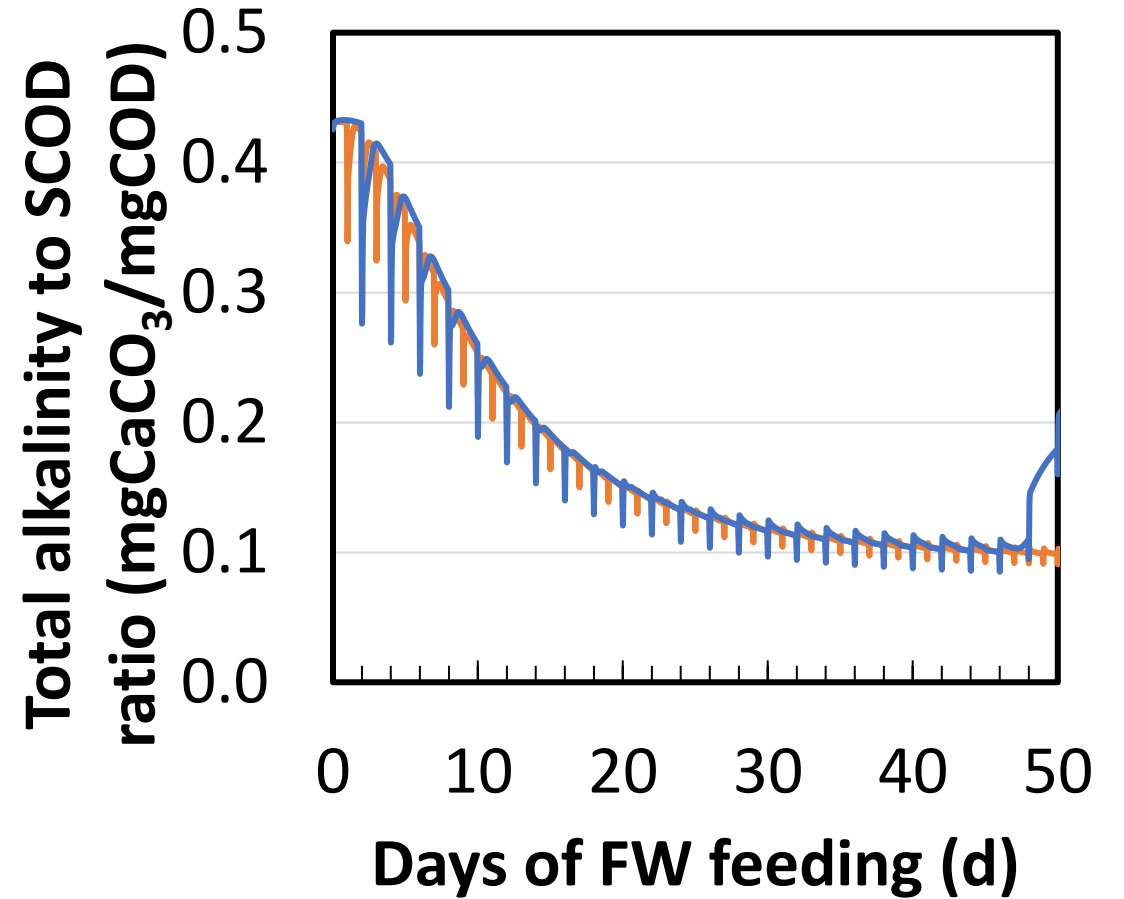
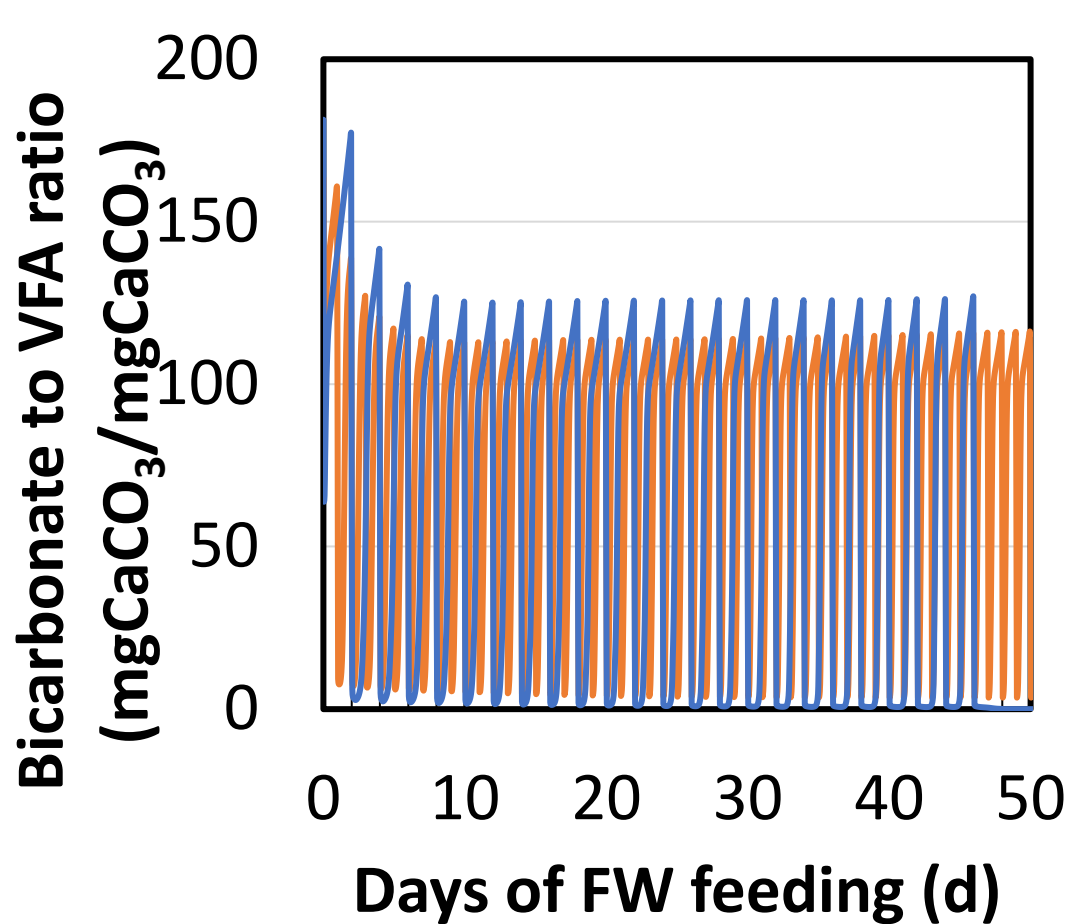
— Once per day — Every 2 days

Other potential indicators were lagging



— Once per day — Every 2 days

Other potential indicators were lagging

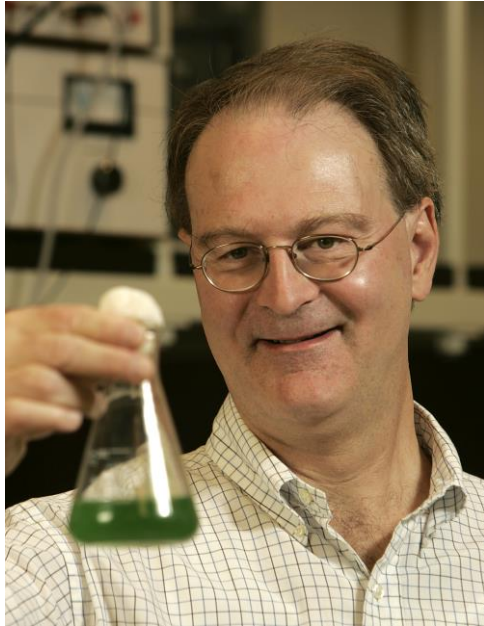


— Once per day — Every 2 days

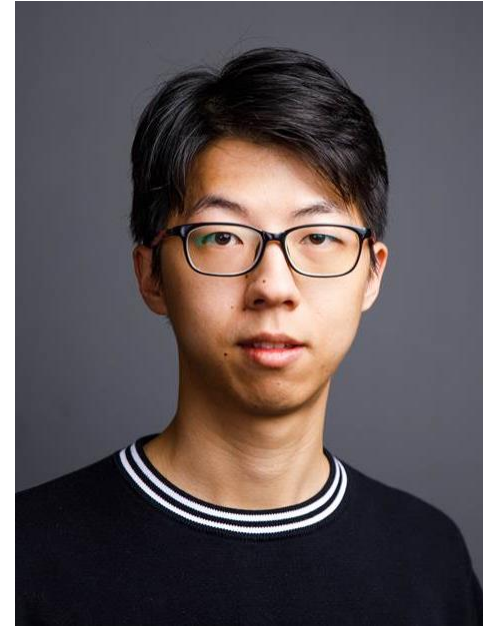
In summary

- Mathematical modeling can be used to predict souring in FW anaerobic co-digestion
- All souring was caused by a depletion of bicarbonate alkalinity
- Bicarbonate alkalinity can be monitored as the leading indicator for souring
 - Bicarbonate alkalinity = total alkalinity - VFA alkalinity

Acknowledgements



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