

IWA Wastewater

Water and Resource Recovery Conference
10 – 13 April 2022, Poznan, Poland



A Thermodynamic Analysis of Intermediary Metabolic Steps and Nitrous Oxide Production in Ammonium-Oxidizing Bacteria

Michelle N. Young, Joshua P. Boltz, Andrew Marcus, Jose A. Jimenez, Ahmed Al-Omari, Imre Takács, and Bruce E. Rittmann



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- Dr. Bruce E. Rittmann
- Dr. Joshua P. Boltz
- Dr. Andrew Marcus



- Dr. Jose A. Jimenez
- Ahmed Al-Omari

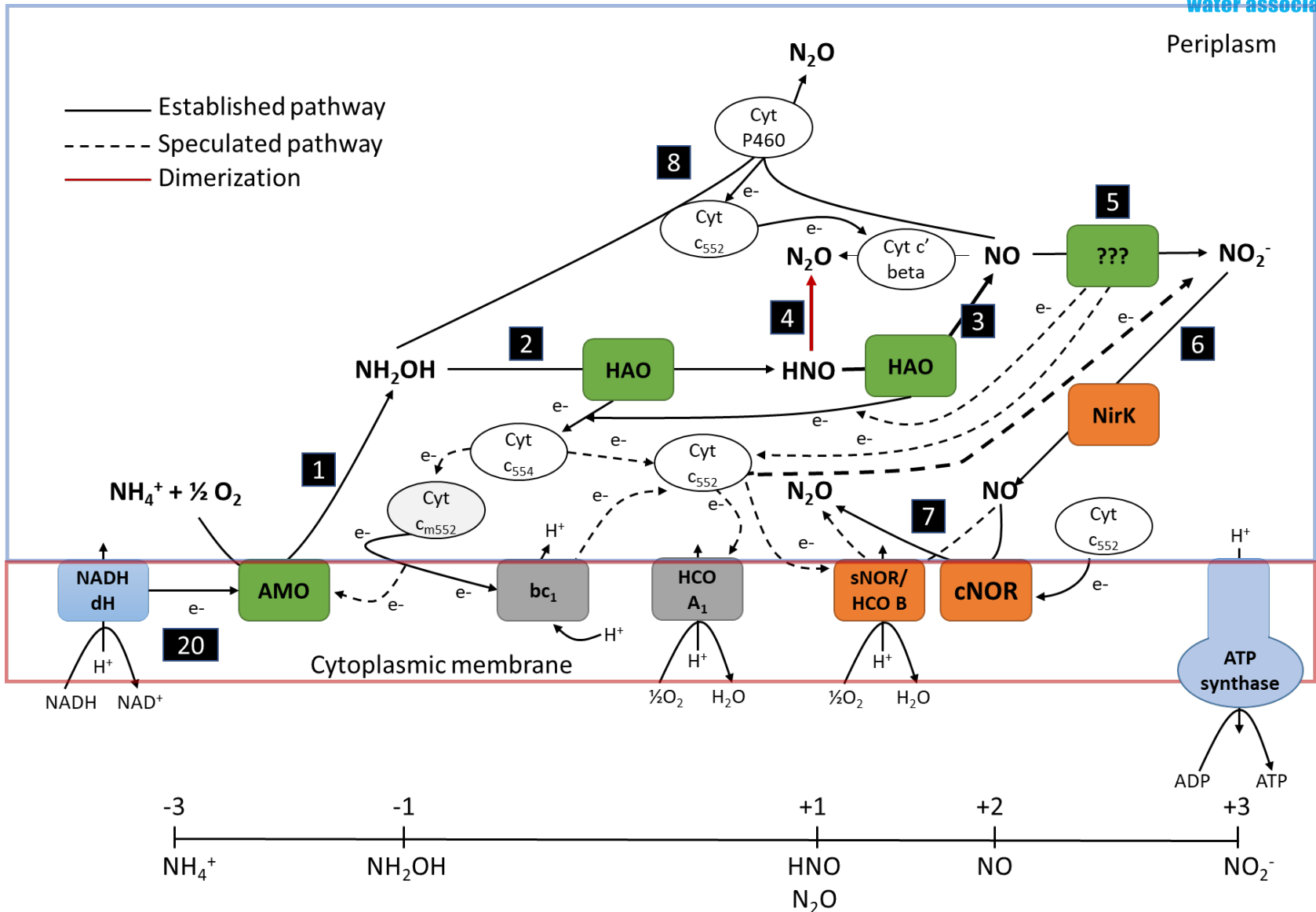


- Dr. Imre Takács

PROJECT GOAL: WORK TOWARDS A BETTER PREDICTIVE MODEL OF AMMONIUM OXIDIZING BACTERIA (AOB) PERFORMANCE

- Gain a better understanding of AOB metabolism to develop a more-accurate model of their performance
- Use metabolic information for better estimates of AOB energetics and kinetics

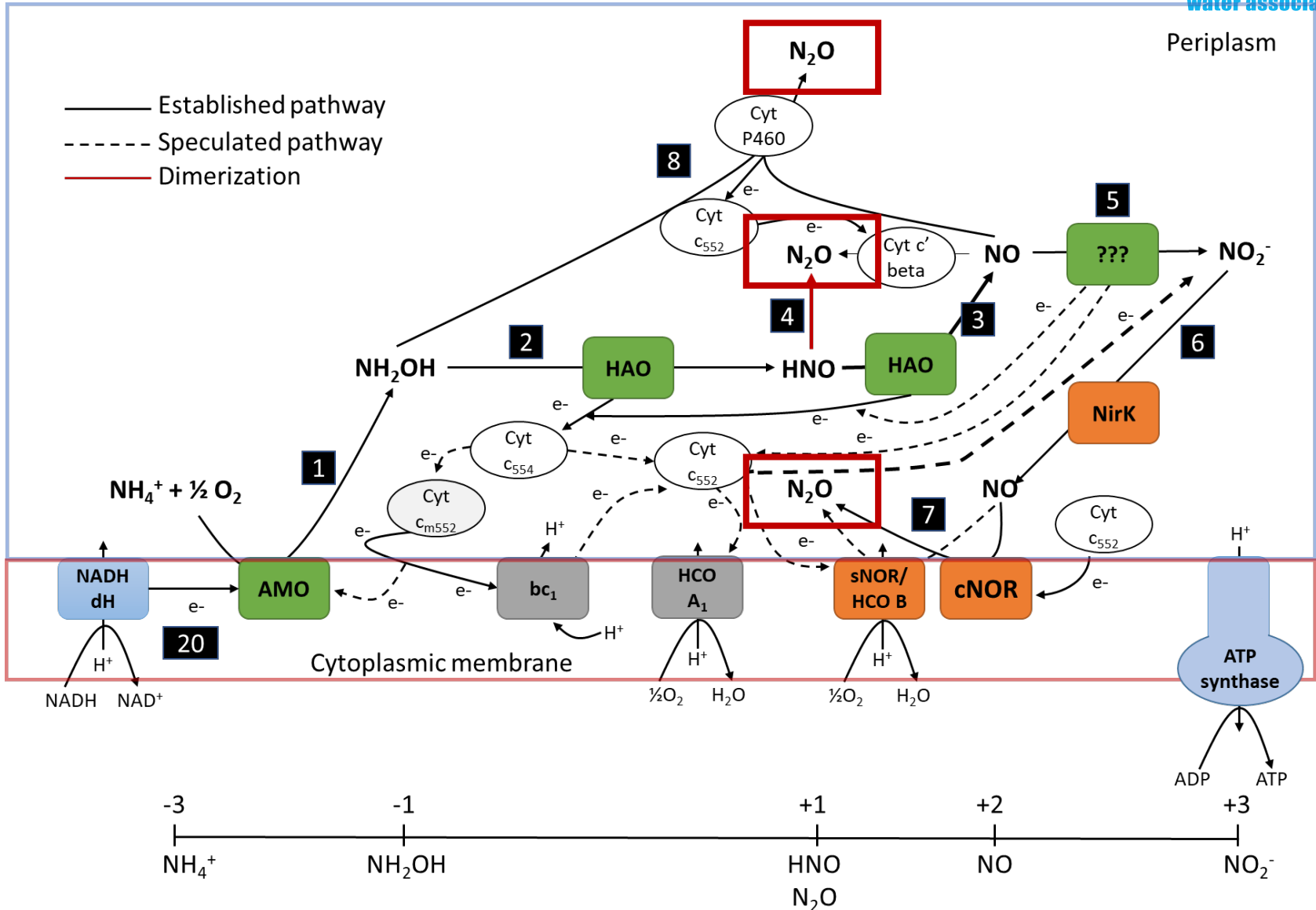
LATEST UNDERSTANDING OF AOB METABOLIC PATHWAY



THREE MISUNDERSTOOD/IGNORED MECHANISMS

- **N_2O can be produced at a variety of steps**
- Ammonium monooxygenation (AMO) is an electron sink
- Respiration and biomass synthesis are not associated with all steps in nitrification or reduction

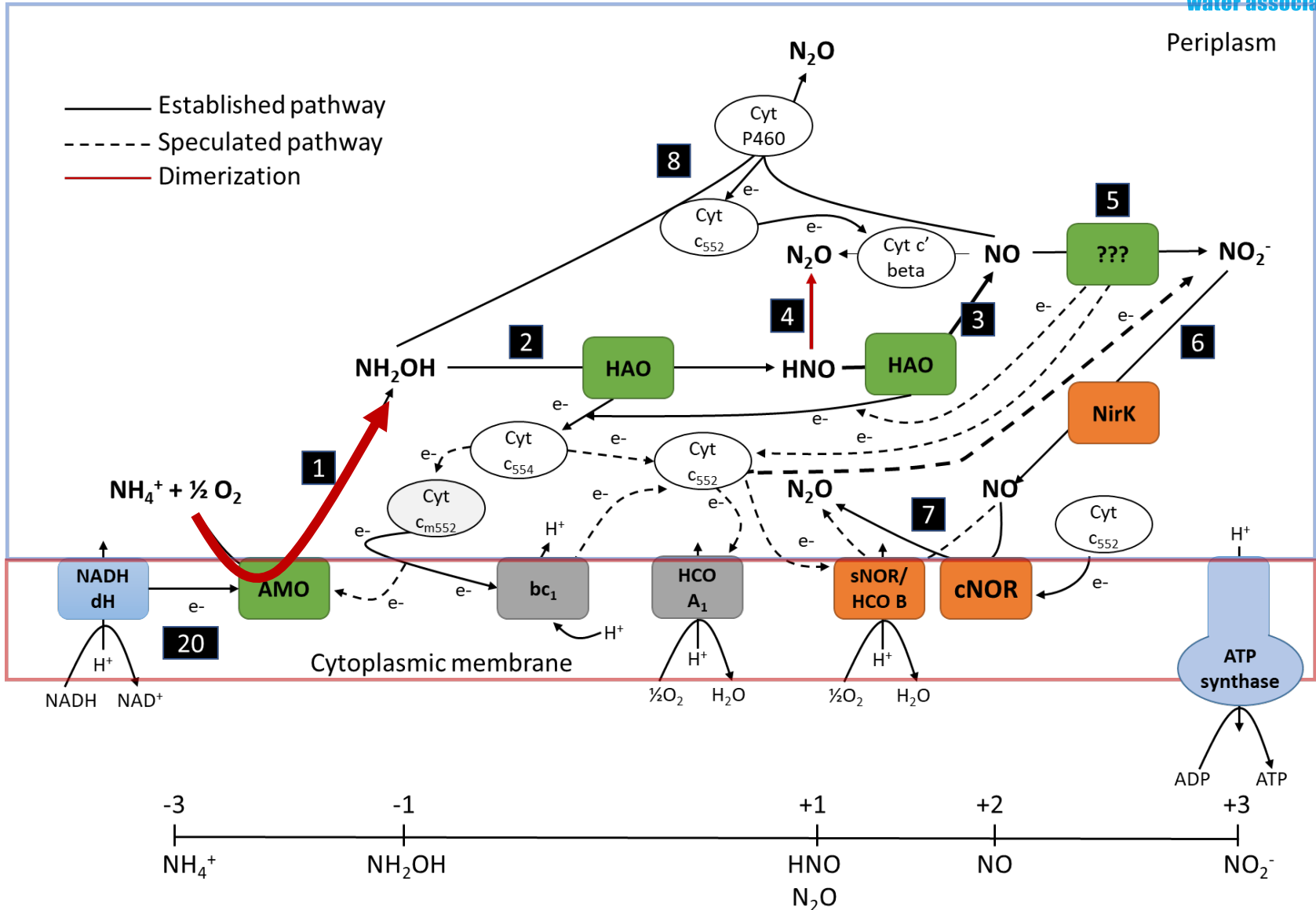
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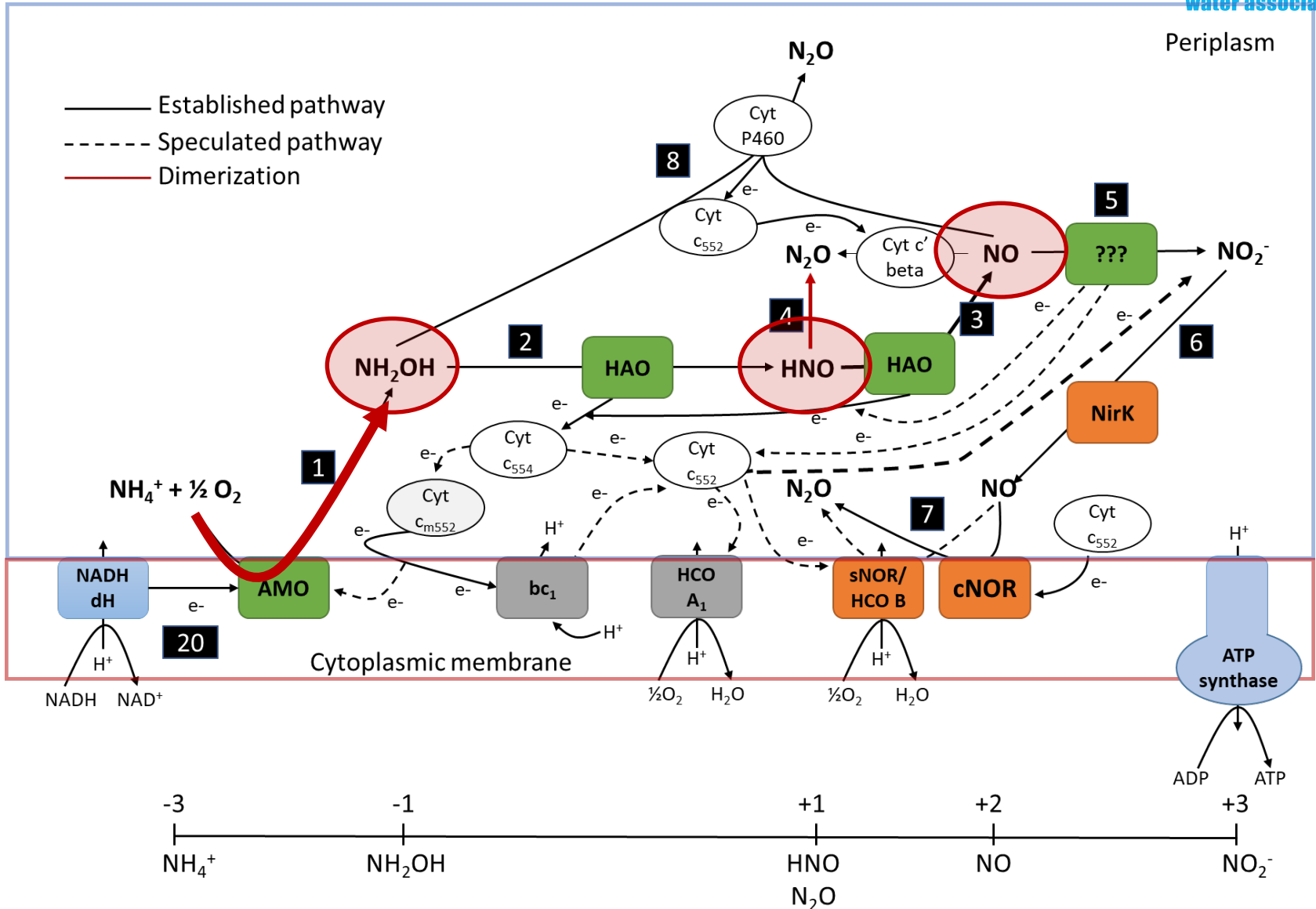
THREE MISUNDERSTOOD/IGNORED MECHANISMS



where [H] = electron equivalent

- 4 out of 6 electrons produced during nitrification are invested in this step!
- 2 additional electrons are from downstream products
- Those electrons do not produce energy or biomass synthesis

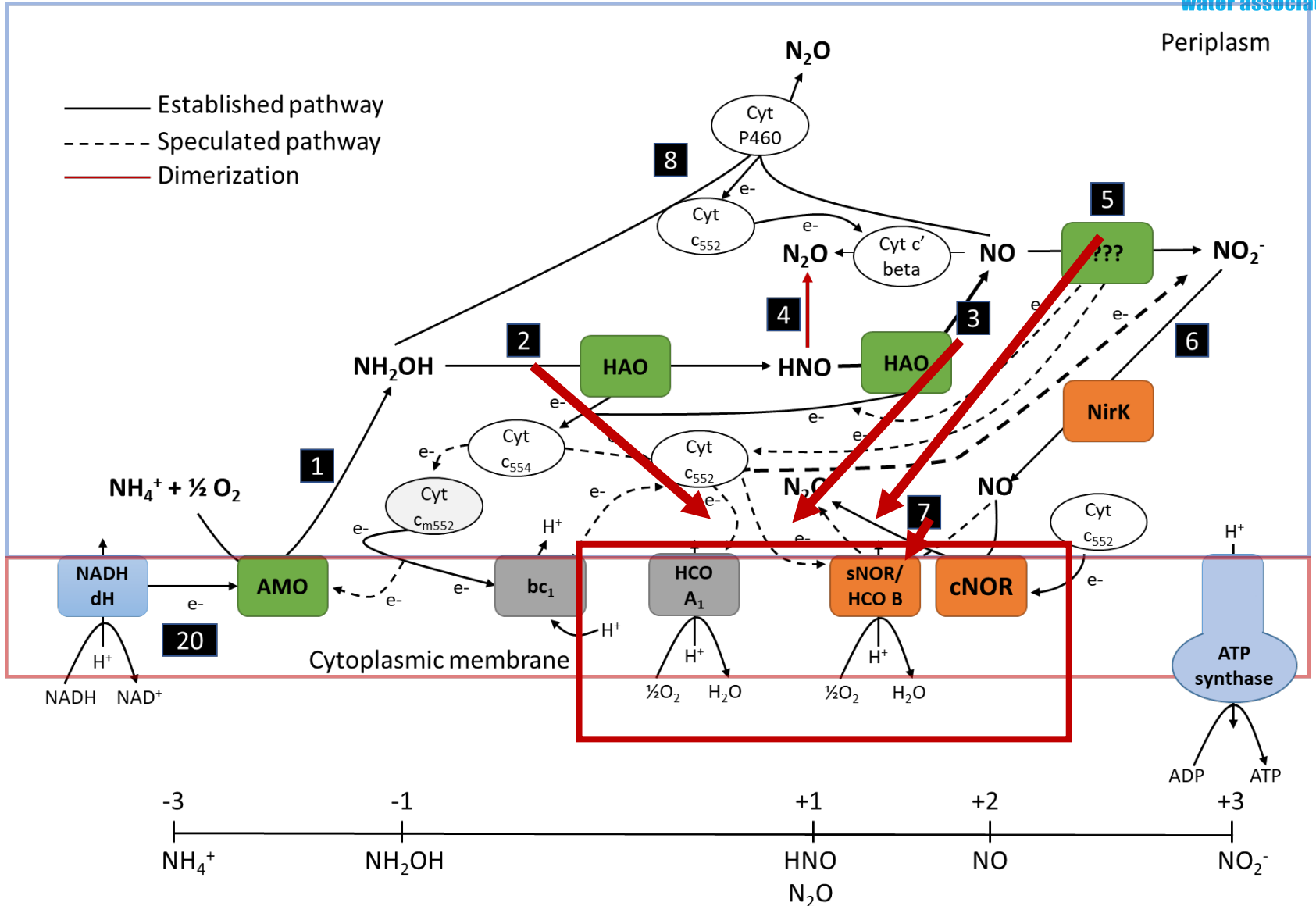
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LATEST UNDERSTANDING OF AOB METABOLIC PATHWAY



ENERGETICS, STOICHIOMETRY, AND KINETICS

- Use the Thermodynamic Electron Equivalents Model (TEEM) to estimate the stoichiometry and kinetics of respiration/biomass growth steps
- Using Gibbs free energies of donor and acceptor half reactions and carbon assimilation energetics, calculate fraction of electrons used for biomass synthesis (f_s^0)
- Using f_s^0 , calculate yield (Y), maximum substrate utilization rate (\hat{q}), maximum specific growth rate ($\hat{\mu}$)
- Described in McCarty (2008) and Sections 5.4 and 6.2 of Rittmann and McCarty (2020)

HYBRID TEEM PATHWAY ANALYSIS

- Calculate the stoichiometry/kinetics for each individual pathway that can produce biomass
- Normalize the individual pathway value based on the number of electrons contributed to overall biomass growth for the whole cell
- Compare whole cell values to literature

INDIVIDUAL PATHWAYS IN NITRIFICATION

Reaction #	Electron donor for respiration and biomass synthesis	Pathway fraction of electron to cell synthesis, f_s^0	Yield, Y (gVSS/gN)	Maximum substrate utilization rate, \hat{q} (g N/gVSS/d)	Maximum specific growth rate, $\hat{\mu}$ (1/d)
1a-d	NH_4^+ to NH_2OH	0	N/A	N/A	N/A
2	NH_2OH to HNO	0.17	0.14	8.4	1.2
3	HNO to NO	0.23	0.19	18.2	1.8
5	NO to NO_2^-	0.24	0.19	18.4	1.8

WHOLE CELL KINETICS FOR NITRIFICATION BASED ON INDIVIDUAL PATHWAY KINETICS

Electron source for the AMO reaction			$f_{s,tot}^0$	Y_{tot}	\hat{q}_{tot}	$\hat{\mu}_{tot}$
NH ₂ OH	HNO	NO	eeq cells/eeq donor	mg X _a /mg N	mg N/ mg X _a /d	1/d
0	1	1	0.057	0.14	3.7	0.70
1	1	0	0.067	0.16	5.0	0.94
1	0	1	0.067	0.16	5.0	0.94
2	0	0	0.078	0.19	7.5	1.42

For comparison, literature values are: 0.15-0.44 3.7-7.5 0.7-1.4

WHOLE CELL KINETICS FOR REDUCTION BASED ON INDIVIDUAL PATHWAY KINETICS

Electron source for the NOR reaction	$f_{s,tot}^0$ eeq cells/eeq donor	Y_{tot} mg X_a /mg N	\hat{q}_{tot} mg N/ mg X_a /d	$\hat{\mu}_{tot}$ 1/d
NH ₂ OH	0.12	0.10	8.0	0.77
HNO	0.15	0.12	8.2	0.96
Literature values based on mathematical modeling only are:		0.02	1.3	0.15-0.28

CONCLUSIONS

- Provide an up-to-date synopsis of AOB metabolism
- Which pathways contribute to biomass growth
- Determine stoichiometry and kinetics based on thermodynamic electron equivalents modeling for the individual pathways
- Reconcile individual pathway values vs whole cell values
- Gained insights into which electron donors are likely preferable under different conditions

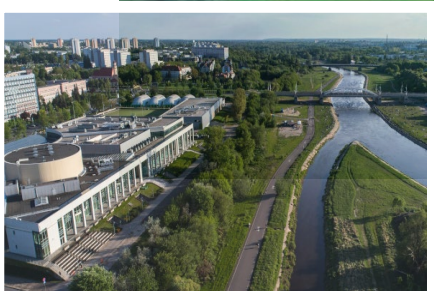
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